



**AAU**

**KING ABDULAZIZ UNIVERSITY  
ACADEMIC ASSESSMENT UNIT**

# **COURSE PORTFOLIO**

**FACULTY OF SCIENCE**

**PHYSICS DEPARTMENT**

**COURSE NAME:** *Solid State Physics*

**COURSE NUMBER:** *Phys 471*

**SEMESTER/YEAR:** *2<sup>nd</sup> / 1436 H*

**DATE:** *Rabia Althani -1437 H*

## ACADEMIC ASSESSMENT UNIT

### Instructor Information

✍ **Name of the instructor:** Dr. Hala A Al-Jawhari

✍ **Office location:** Faculty of Science, 3<sup>rd</sup> floor, Room No. (108).

✍ **Office hours:**

Sunday	Monday	Tuesday	Wednesday	Thursdays
10-11, 12-1	-	10-11, 12-1	-	10-11, 11-2

✍ **Contact number(s):** Tel : 6400000 Ext. 26904

✍ **E-mail address:** haljawhari@kau.edu.sa

### Course Information

✍ **Course name and number:** Solid State Physics - Phys (471)

✍ **Course meeting times & places:**

	Time	Room
Lectures	9-10 Sun, Tus & Thru.	104A

✍ **Course prerequisites and requirements:** Quantum Mechanics **Phys (354)**

✍ **Description of the course :**

This course is an introduction to the: crystal structures, properties of periodic lattices, electrons in metals, band structure, transport properties, semiconductors, and superconductivity.

### Course Objectives

By the end of this course student must be able to:

- Define the lattice planes & directions.
- Establish a reciprocal lattice from a real lattice.
- Determine *the structure factor* for all cubic lattices.
- Define phonons in crystals and distinguish between their different modes.
- Choose the right formulas to calculate *specific heat* of the lattice.
- Recognize the main drawbacks of the *free electron model* in metals.
- Identify: Bloch's theorem, Brillouin zones & Fermi surface in metals.
- Classify different types of solid according to *The Band Theory*.
- Recognize the main factors affect *the conductivity in metals*.
- Distinguish between *intrinsic & extrinsic Semiconductors* and know their properties and applications.
- Recognize the idea behind the Superconductivity phenomenon and be aware of its applications.

### Learning Resources

✍ The primary textbook for the course is

**1- Elementary Solid State Physics by M. Ali Omar, 1997.**

✍ Additional references that may be helpful to students in this course:

## ACADEMIC ASSESSMENT UNIT

- 2- *Solid State Physics* , by J. Hook & H. Hall , 2<sup>nd</sup> ed (2010).
- 3- *Materials Science & Engineering*, by W. Callister & D. Rethwisch, 8<sup>th</sup> ed (2011).
- 4- *The Physics of Solid*, by R. Turton, (2000).
- 5- *Introduction to Solid State Physics*, by C. Kittel, 6<sup>th</sup> ed .
- 6- <http://www.doitpoms.ac.uk/tlplib/index.php>
- 7- <http://ocw.mit.edu/courses/materials-science-and-engineering/3-091sc-introduction-to-solid-state-chemistry-fall-2010/index.htm>

### Course Requirements and Grading

#### Student assessment:

H.W	10%	-
Quiz	10%	The 2 <sup>nd</sup> lecture of each week
1 <sup>st</sup> EXAM	20%	21-5-1437 H
2 <sup>nd</sup> EXAM	20%	5-7-1437 H
Final Exam	40%	After week 14

#### Expectations from students:

- Students are expected to attend lectures and labs on time.
- Students may discuss a homework assignment to clarify what is required. However, students may NOT share or copy each other work.
- Students are expected to prepare them self for each exam or quiz, and bring their own pens and calculator.

#### Student responsibilities to the course:

The student responsibilities are to:

- Attend all lectures and labs, since the absence of 25% of the total lectures will prevent the student from attending the final exam.
- Perform all lab experiments and submit their required reports.
- Submit each problem set on its defined date. .
- Work hard on each given exam or quiz. **Eight** quizzes will be given, but only the best **five** of them will be encountered.

### Detailed Course Schedule

	Topic	Readings	Due
1 (14/4/37)	<b>STRUCTURE &amp; BONDING</b> Periodic Structures & main definitions.	Ch 1, Sec 1-3	
	Crystal Planes & directions	Ch 1, Sec 4-6	+Ch 3 -Ref(3)

## ACADEMIC ASSESSMENT UNIT

	Common Structures 1	Ch 1, Sec 7	
2	Common Structures 2	Ch 1, Sec 7	
	Interatomic Forces	Ch 1, Sec 9	
	Types of Bonding	Ch 1, Sec 10	
3	<b>DIFFRACTION IN CRYSTALS</b> X-Ray (Nature & Generation) & Bragg's Law.	Ch 2, Sec 1-4	
	Diffraction by a Crystal	Ch 2, Sec 5	
	Reciprocal Lattice & Diffraction	Ch 2, Sec 6-7	
4	The structure factor.	Ch 2, Sec 9-11	
	Neutron and Electron Diffraction.	Ch 2, Sec 12	
	<b>Tutorial</b> session 1		<b>Problem set (1)</b>
5	<b>ACCOUSTIC PROPERTIES OF SOLIDS</b> The Continuum Model(1D) & Discrete Lattice	Ch 3, Sec 1-2 & 6	
	Dispersion Relation for a 2D Lattice & Phonons	Ch 3, Sec 5-6	
	Density of States for a 1D & 3D Lattice	Ch 3, Sec 3,7	
<b>*** 1<sup>st</sup> EXAM *** 21-5-1437</b>			
6	Specific Heat	Ch 3, Sec 4,8	
	Specific Heat & Thermal Conductivity	Ch 3, Sec 4,8 & 9	
	<b>THE FREE ELECTRON MODEL</b> Drude Model of metals	Ch 4, Sec 1-4	
7	Electrical conductivity vs temperature	Ch 4, Sec 5	
	The Quantum Model & the Fermi Sphere	Ch 4, Sec 6-8	
	Thermal conductivity & Hall Effect	Ch 4, Sec 9-10	
<b>Mid Term Holiday</b>			
8 (11/6/37)	<b>Tutorial</b> session 2		<b>Problem set (2)</b>
	<b>BAND THEORY OF SOLIDS</b> Formation of Energy Bands: Bloch Theorem	Ch 5, Sec 1-3	
	The crystal potential	Ch 5, Sec 5	
9	No. of states in a band	Ch 5, Sec 5	
	The Nearly Free Electron Model	Ch 5, Sec 6-7	
	Metals, Insulators, & Semiconductors- DOS	Ch 5, Sec 10 -11	
<b>*** 2<sup>nd</sup> EXAM *** 5-7-1437</b>			
10	Fermi surface & velocity	Ch 5, Sec 12-13	
	The electron effective mass	Ch 5, Sec 14-16	
	The Hole, Electrical conductivity & Hall Effect	Ch 5, Sec 17-19	
11	<b>SEMICONDUCTORS</b> Intrinsic semiconductors	Ch 6, Sec 1-4	
	Doped semiconductors	Ch 6, Sec 5-6	
	Cont.		
12	Mobility & The Semiconductor Hall Effect	Ch 6, Sec 7-8	
	The hot electrons & Gunn Effect	Ch 6, Sec 10-11	
	Optical Absorption	Ch 6, Sec 12	
13	<b>Tutorial</b> session 3		<b>Problem set (3)</b>

## ACADEMIC ASSESSMENT UNIT

	<b><i>SUPERCONDUCTIVITY</i></b> <i>Types I &amp; II Superconductors &amp; The BCS Theory</i>	<i>given Handout</i>	
	<i>High T<sub>c</sub> Superconductors &amp; their applications</i>	<i>given Handout</i>	
<b>14</b>	<b>*** FINAL EXAM ***</b>		